

Chapter 19

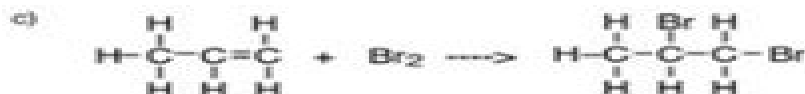
1. a) Contains as many hydrogen atoms as possible for the given number of carbons. Only carbon-carbon single bonds.



(ii) liquid



2. a) a carbon-carbon double bond



(Home schooling parents: It doesn't matter what position you draw the bromine in on the right-hand carbon - N, S or E (with hydrogens in the other two positions). All that matters is that one bromine becomes attached to each of the carbons either side of the original double bond.)



(ii) In this (substitution) reaction, one of the hydrogens has been replaced by a bromine during the reaction. In the previous (addition) reaction, nothing was lost when the two molecules combined together.

3. a) Ethene is passed with steam at 300°C and 60 - 70 atmospheres over a phosphoric acid catalyst.

b) e.g. two of: continuous flow process, therefore faster; rapid reaction; purer product.

c) uses renewal resources; gentle conditions.

4. Dissolve the sugar in water in a conical flask (or similar) and add yeast. You now need some means of keeping the air out while allowing carbon dioxide to escape. You could use a commercial home-winemaking airlock of some sort, but a plug of cotton wool works just as well. Leave the mixture in the warm for a few days until bubbling stops.



Allow the yeast to settle, and pour off the solution so that it can be fractionally distilled, collecting the fraction which boils at 78°C . (See the diagram for the fractional distillation of an alcohol/water mixture in Chapter 11 of the book.)

continued . . .

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Brendan G. Carr



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Table of Contents Catalyst Lab Manual For Chemistry 122 Answers

1. Understanding the eBook Catalyst Lab Manual For Chemistry 122 Answers
 - The Rise of Digital Reading Catalyst Lab Manual For Chemistry 122 Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Catalyst Lab Manual For Chemistry 122 Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Catalyst Lab Manual For Chemistry 122 Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Catalyst Lab Manual For Chemistry 122 Answers
 - Personalized Recommendations
 - Catalyst Lab Manual For Chemistry 122 Answers User Reviews and Ratings
 - Catalyst Lab Manual For Chemistry 122 Answers and Bestseller Lists

5. Accessing Catalyst Lab Manual For Chemistry 122 Answers Free and Paid eBooks
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 - Catalyst Lab Manual For Chemistry 122 Answers eBook Subscription Services
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6. Navigating Catalyst Lab Manual For Chemistry 122 Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Catalyst Lab Manual For Chemistry 122 Answers Compatibility with Devices
 - Catalyst Lab Manual For Chemistry 122 Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Catalyst Lab Manual For Chemistry 122 Answers
 - Highlighting and Note-Taking Catalyst Lab Manual For Chemistry 122 Answers
 - Interactive Elements Catalyst Lab Manual For Chemistry 122 Answers
8. Staying Engaged with Catalyst Lab Manual For Chemistry 122 Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Catalyst Lab Manual For Chemistry 122 Answers
9. Balancing eBooks and Physical Books Catalyst Lab Manual For Chemistry 122 Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Catalyst Lab Manual For Chemistry 122 Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Catalyst Lab Manual For Chemistry 122 Answers
 - Setting Reading Goals Catalyst Lab Manual For Chemistry 122 Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Catalyst Lab Manual For Chemistry 122 Answers
 - Fact-Checking eBook Content of Catalyst Lab Manual For Chemistry 122 Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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