

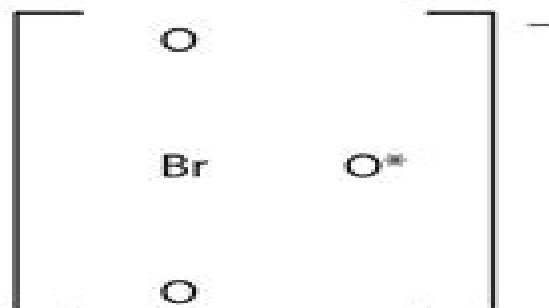
19 Potassium bromate(V), KBrO_3 , is a primary standard, meaning that it can be obtained as a pure substance and used to accurately determine the concentrations of solutions of other chemicals, such as sodium thiosulfate, $\text{Na}_2\text{S}_2\text{O}_3$.

- (a) (i) Complete the dot and cross diagram for the bromate(V) ion. Show only the outer shell electrons.

In this ion, the bromine expands its outer shell to accommodate 12 electrons.

Use $\times$ for bromine electrons and $\bullet$ for oxygen electrons. The symbol * on the diagram represents the extra electron which gives the ion its charge.

(2)



- (ii) Suggest how elements in Period 3 and higher can accommodate more than eight electrons in their outer shell.

(1)



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