

IONIC AND METALLIC BONDING

Chapter Test A

A. Matching

Match each description in Column B with the correct term in Column A. Write the letter of the correct description on the line.

| | Column A | Column B |
|---|---------------------------|---|
| E | 1. electron dot structure | a. a mixture of two or more elements, at least one of which is a metal |
| F | 2. ionic compound | the number of ions of opposite charge surrounding each ion in a crystal |
| H | 3. valence electron | the force of attraction binding oppositely charged ions together |
| C | 4. ionic bond | d. the attraction of valence electrons for positive metal ions |
| J | 5. chemical formula | a depiction of valence electrons around the symbol of an element |
| | | f. compound of cations and anions |
| G | 6. halide ion | g. an anion of a halogen |
| Α | 7. alloy | h. an electron in the highest occupied energy level of an atom |
| 1 | 8. octet rule | Atoms in most compounds tend to achieve the electron configuration of a noble gas. |
| K | 9. formula unit | |
| | | shows the kinds and numbers of atoms in the smallest representative unit of a substance |
| В | 10. coordination number | k. lowest whole-number ratio of ions in an ionic compound |
| D | 11. metallic bond | ac sowest whose-manner ratio of some in an some compound |

B. Multiple Choice

Choose the best answer and write its letter on the line.

| 12 | . How many | valence electr | ons does an at | om of any el- | ement in Group 6 | A. harve? |
|--------|--|----------------|----------------|---------------|------------------|-----------|
| | the state of the s | | | | | |

a. 2 c. 6 b. 4 d. 8

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Thomas Bond, Chris Hughes

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