

Chemistry — Unit 8 Test

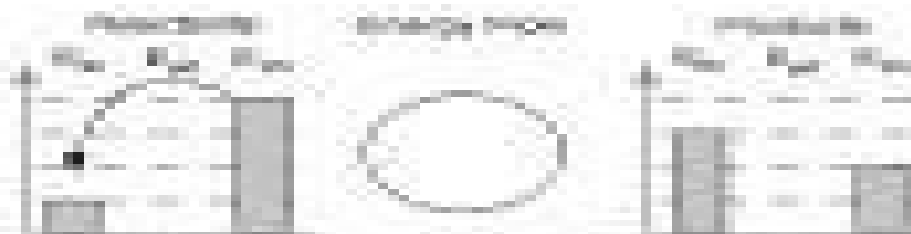
1. In this ideal gas law _____ represents number of gas _____
 n V P T

2. The activation energy _____
- The activation energy decreases when the products _____
 - The products have more energy than the reactants _____
 - The kinetic energy of the system decreases _____
 - It is the position in position _____

3. Consider the energy level _____

Which of the following _____
 state about the _____
 reaction is FALSE?

- The reaction is exothermic _____
- The products have less _____
 energy than the reactants _____
- The temperature of the _____
 system decreases _____
- The energy level _____
 is higher than the energy level _____



Match volume and gas solubility

4. The three gas solubility _____
 are _____, _____, and _____
 Match the gas solubility _____
 with the volume _____

5. The gas solubility _____
 is _____

6. When potassium chlorate _____
 is decomposed, oxygen gas and potassium chloride _____
 are produced. What volume of oxygen gas at STP is produced when 2.75 g of KClO₃ _____
 decomposes? _____
 Answer: _____

Chemistry Unit 9 Test

Judd E. Hollander



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