

3 • Organic Chemistry**UNIT TEST • PRACTICE**

- You may use a pencil, eraser, and scientific calculator to complete the test. No other resources are allowed.
- You have fifty (50) minutes to complete this test, unless other arrangements have been made.
- Please transfer your answers for Matching and Multiple Choice questions onto the Answer Document. Work for the Matching and Multiple Choice questions will not be graded. However, sufficient and appropriate work must be shown for the Free Response questions in order to receive full credit.

SECTION 1: MATCHING. Use the following answer choices that best match each description. Answer choices may be used once, more than once, or not at all. (1 point each)

Questions 1-6:

- (A) Alkanes only
- (B) Alkenes only
- (C) Alkynes only
- (D) Both alkenes and alkynes

1. Have a molecular formula C_4H_8 .
2. Have *cis-trans*- isomers.
3. Can rotate around C–C bond.
4. Have a triple bond in the main carbon chain.
5. Have an H–C–C bond angle of 120° .
6. C_4H_8 is an example.
7. Are considered “saturated” hydrocarbons.

Questions 7-12:

- (A) Ketone
- (B) Amine
- (C) Carboxylic Acid
- (D) Ether

8. Has a carbon bonded to a nitrogen.
9. Has a sour taste.
10. Acetone, or nail polish remover, is an example.
11. Has a carbon that is bonded to oxygen only with a double bond.
12. Has an oxygen single bonded to two carbons.
13. Vinegar is an example.

SECTION 2: MULTIPLE CHOICE. Select the best answer choice for each question. (2 points each)

14. Which of the following elements has a bonding capacity of 2?
 - (A) Oxygen
 - (B) Hydrogen
 - (C) Carbon
 - (D) Chlorine

15. Structural formulas differ from molecular formulas in that they show
 - (A) The arrangement of atoms.
 - (B) The bond angles in the molecule.
 - (C) The elements that make up the compound.
 - (D) The number of each element present.

Chemistry Unit Test Practice

Arshad Iqbal



Chemistry Unit Test Practice:

The book delves into Chemistry Unit Test Practice. Chemistry Unit Test Practice is a crucial topic that needs to be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Chemistry Unit Test Practice, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to Chemistry Unit Test Practice
- Chapter 2: Essential Elements of Chemistry Unit Test Practice
- Chapter 3: Chemistry Unit Test Practice in Everyday Life
- Chapter 4: Chemistry Unit Test Practice in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, the author will provide an overview of Chemistry Unit Test Practice. The first chapter will explore what Chemistry Unit Test Practice is, why Chemistry Unit Test Practice is vital, and how to effectively learn about Chemistry Unit Test Practice.
3. In chapter 2, this book will delve into the foundational concepts of Chemistry Unit Test Practice. The second chapter will elucidate the essential principles that must be understood to grasp Chemistry Unit Test Practice in its entirety.
4. In chapter 3, this book will examine the practical applications of Chemistry Unit Test Practice in daily life. This chapter will showcase real-world examples of how Chemistry Unit Test Practice can be effectively utilized in everyday scenarios.
5. In chapter 4, the author will scrutinize the relevance of Chemistry Unit Test Practice in specific contexts. This chapter will explore how Chemistry Unit Test Practice is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, the author will draw a conclusion about Chemistry Unit Test Practice. The final chapter will summarize the key points that have been discussed throughout the book.

This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Chemistry Unit Test Practice.

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