

The Mole Study Guide

Mass Composition: What's your element's % of mass after the reaction? Molar mass to obtain your value!

10. There are 3.75 moles of CO_2 .

11. There are _____ moles of _____.

12. There are _____ moles of _____.

13. There are _____ moles of _____.

14. There are 2.5 moles of _____.

15. There are _____ moles of _____.

16. There are _____ moles of _____.

17. There are _____ moles of _____.

18. There are 4.5 moles of _____.

19. There are _____ moles of _____.

20. There are _____ moles of _____.

21. There are 3.5 moles of _____.

22. There are _____ moles of _____.

23. There are _____ moles of _____.

24. There are 2.5 moles of _____.

25. There are _____ moles of _____.

26. There are _____ moles of _____.

Name _____ Period _____ Date _____

The Mole Study Guide

Fill in the Blank boxes for the image below.



Write the Empirical Formula for each compound. Use subscripts to show the relative ratio of atoms.

Molecular	Empirical
$\text{C}_6\text{H}_{12}\text{O}_6$	
$\text{C}_4\text{H}_8\text{O}_4$	
$\text{C}_2\text{H}_4\text{O}_2$	
$\text{C}_3\text{H}_6\text{O}_3$	
$\text{C}_2\text{H}_4\text{O}_2$	

Calculate the Molar Mass for the following compounds.

10. H_2O

11. H_2O_2

12. H_2O

13. H_2O_2

Calculate the Percent Composition for each element in the following compound.

10. H_2O

11. H_2O_2

12. H_2O

13. H_2O_2

14. H_2O

15. H_2O_2

16. H_2O

17. H_2O_2

18. H_2O

19. H_2O_2

Determine the Empirical and Molecular Formula for the compound given the information below.

10. A compound has the following percent composition: 58.80% Hg, 21.80% Cl, 1.40% N. The molar mass for Aqueous is 100.0 g/mol.

a. If you have 100 g of Aqueous, how many grams of Hg, Cl, and N do you have?

b. Determine the Empirical Formula for Aqueous.

c. Determine the Molecular Formula for Aqueous.

Chemistry The Mole Study Guide Answers

Scott C. Dulebohn



Chemistry The Mole Study Guide Answers:

The Enigmatic Realm of **Chemistry The Mole Study Guide Answers**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing short of extraordinary. Within the captivating pages of **Chemistry The Mole Study Guide Answers** a literary masterpiece penned by a renowned author, readers attempt a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting impact on the hearts and minds of those that partake in its reading experience.

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