

ChemQuest 1.1

Intro. to Chemical Reactions

Name: _____
 Date: _____
 Class: _____

Electrolyte Review

1. Write the formula for the following compounds:

- a) barium phosphate: $\text{Ba}_3(\text{PO}_4)_2$ b) sodium phosphate: Na_3PO_4
 c) magnesium phosphate: $\text{Mg}_3(\text{PO}_4)_2$ d) lithium nitrate: LiNO_3

2. Write the name for the following compounds:

- a) $\text{Cu}(\text{NO}_3)_2$: copper(II) nitrate b) FeCl_3 : iron(III) chloride
 c) SnSO_4 : tin(II) sulfate d) SnF_4 : tin(IV) fluoride

Chemistry 1 Experiment: What does it do?

There is what a chemical reaction looks like when it is written:



There are examples that you can use to describe what happens during:

Chemistry 1 results in the reaction and the reaction and the reaction.

There are also some characteristics that you can use to describe the reaction. Here are a few examples:

Critical Thinking Questions

1. Write the following reaction, with the reactants and products:



Chemistry 1 results in the reaction and the reaction and the reaction.

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