

25 Nuclear Chemistry



ELECTRONS AND THE STRUCTURE OF ATOMS

25.1 Nuclear Radiation



For students using the Foundation edition, assign problems 1–7, 9, 12, 13.

Essential Understanding

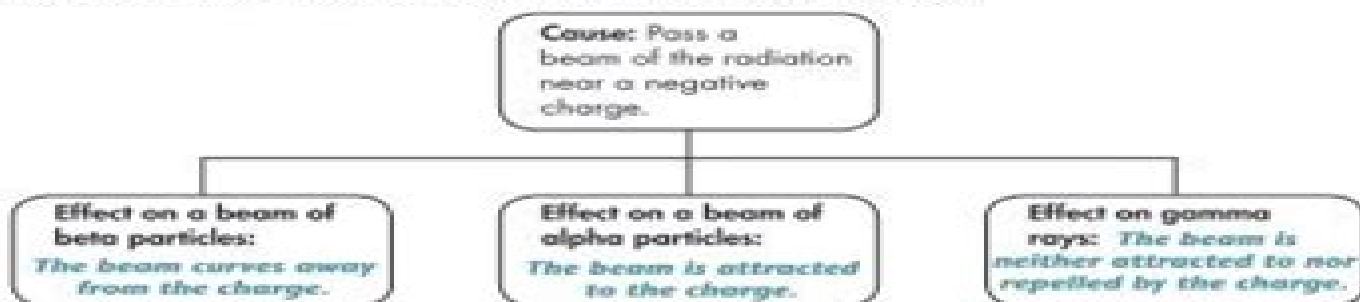
Nuclear reactions might result in the emission of three different types of nuclear radiation: alpha, beta, and gamma.



Reading Strategy

Cause and Effect A cause and effect chart is a useful tool when you want to describe how, when, or why one event causes another. A cause is the reason something happens. The effect is what happens.

As you read Lesson 25.1, use the cause and effect chart below. Complete the chart by filling in the effect of subjecting each type of radiation to a negative charge.



EXTENSION Draw a diagram to illustrate each of the effects.

Diagrams should show a beam of beta particles curving away from a negative charge, a beam of alpha particles curving toward a negative charge, and a gamma ray being unaffected by a negative charge and traveling in a straight path.

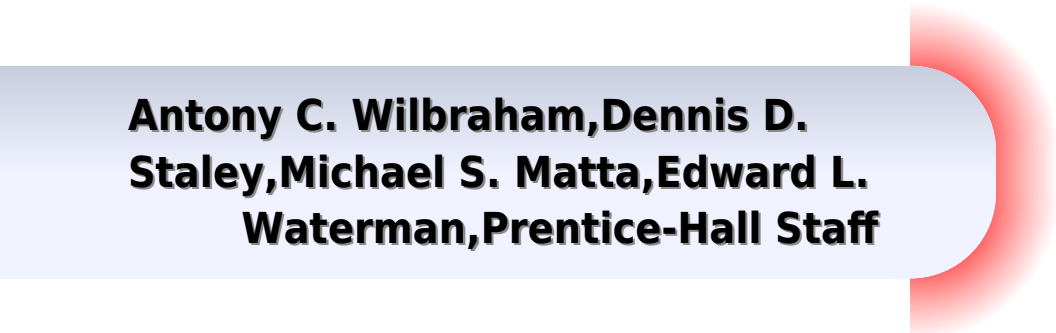
Lesson Summary

Radioactivity Radioactivity is the spontaneous emission of particles or energy from unstable nuclei of radioisotopes.

- The term *radioactivity* refers to the process of emitting particles or energy from atomic nuclei.

Chapter 25 Nuclear Chemistry Guided Reading Answers

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