



11 How does the arrangement of particles in a solid differ from the arrangement of particles in a liquid? What is the change of state called? Can you give examples of substances that can do this?

10 Complete the diagram to show the arrangement of the particles in each state of matter. Label the particles and the change of state.

9 What is a change of state, and what does it mean?

8 What property of matter is demonstrated by the different states of matter? Give examples of the different states of matter.

7 How do the particles in a solid differ from the particles in a liquid?

PARTICLE THEORY AND STATES OF MATTER

Particle theory explains the properties of matter by describing the arrangement of particles in a substance. The particles in a substance are made up of different types of particles, for example, atoms and molecules. The arrangement of particles in a substance determines its properties.

States of Matter: Matter can exist in three primary states: solid, liquid, and gas. In each state, the particles are arranged differently. In a solid, the particles are packed closely together. In a liquid, the particles are more spread out. In a gas, the particles are very spread out and move freely.

Changes of State: Matter can change from one state to another. These changes are called changes of state. Examples of changes of state include melting, freezing, boiling, and condensation. These changes occur because the particles in a substance are moving and interacting with each other.

Particle Theory: Particle theory explains the properties of matter by describing the arrangement of particles in a substance. The particles in a substance are made up of different types of particles, for example, atoms and molecules. The arrangement of particles in a substance determines its properties.



5 What are the three states of matter?

6 How do the particles in a solid differ from the particles in a liquid?

Chapter 13 States Of Matter Guided Reading Answer Key

John K. Lee



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