

**Section 1.1** *continued*

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

|     |
|-----|
| 19  |
| 39  |
| 55  |
| 81  |
| 113 |
| 151 |

|                        |      |
|------------------------|------|
| Atomic Mass            | 19.  |
| Atomic Number          | 39.  |
| Electron Configuration | 55.  |
| Chemical Name          | 81.  |
| Chemical Symbol        | 113. |

Put each item in Column A, either the letter of the matching item in Column B.

**Column A**

- \_\_\_\_\_ 16. It exists in the periodic table.
- \_\_\_\_\_ 17. It has no line periodic table.
- \_\_\_\_\_ 18. Group 1 elements.
- \_\_\_\_\_ 19. Elements that are shiny and conduct electricity.
- \_\_\_\_\_ 20. Group 8 elements.

**Column B**

- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write *yes* if the statement is true, if the statement is false, change the underlined word or phrase to make it true.

- \_\_\_\_\_ 21. There are two representative elements of elements.
- \_\_\_\_\_ 22. Most elements have lines in the periodic table for metalloids.
- \_\_\_\_\_ 23. Group 1A elements except for hydrogen are metals in the solid state.
- \_\_\_\_\_ 24. Group 2A elements are the alkaline earth metals.
- \_\_\_\_\_ 25. Group 7A elements are highly reactive nonmetals known as halogens.
- \_\_\_\_\_ 26. Group 8A elements are inert nonmetallic elements known as noble gases.
- \_\_\_\_\_ 27. Elements have properties of both metals and nonmetals.

# Chapter 12 Study Guide For Content Mastery

## Stoichimetry

**M Walker**



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