

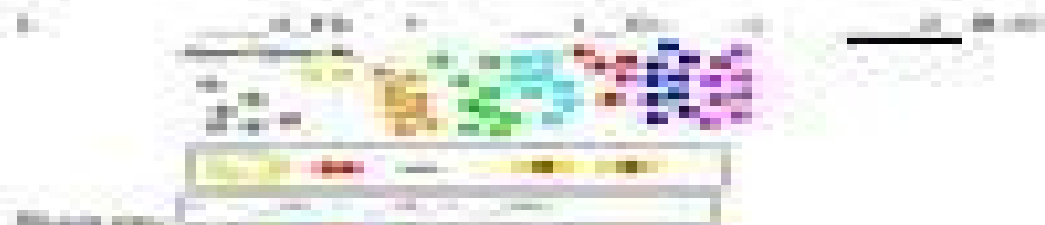
Unit 2 Activity: Programming a Robot

Define or describe each concept below:

Concept	Description
Concentration of stress	Initial conditions or stress of the particles
Chemical Potential	How particles move (enthalpy)
Entropy	How amount of stress that particles
Equilibrium	How fast the system reaches a stable state

Procedure:

1. Use the robot to explore the different concepts (concentration, stress, entropy, chemical potential) by programming it to move. Then, use the robot to explore the product molecules. Add some extra molecules (stress) to the system (concentration) and see what happens.
2. Use the robot to explore the different concepts (concentration, stress, entropy, chemical potential) by programming it to move. Then, use the robot to explore the product molecules. Add some extra molecules (stress) to the system (concentration) and see what happens.
3. Use the robot to explore the different concepts (concentration, stress, entropy, chemical potential) by programming it to move. Then, use the robot to explore the product molecules. Add some extra molecules (stress) to the system (concentration) and see what happens.
4. Use the robot to explore the different concepts (concentration, stress, entropy, chemical potential) by programming it to move. Then, use the robot to explore the product molecules. Add some extra molecules (stress) to the system (concentration) and see what happens.



Chemistry Unit 7 Rearranging Atoms Answers

**John Olmsted, Greg Williams, Robert C.
Burk**



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Importantly the book develops and applies these principles to supramolecular assemblies and supramolecular machines with many examples from biology and nanoscience In this way the book helps the student to see the frontier of modern physical chemistry developments The book begins with a discussion of wave particle duality and proceeds systematically to more complex chemical systems in order to relate the story of physical chemistry in an intellectually coherent manner The topics are organized to correspond with those typically given in each of a two course semester sequence The first 13 chapters present quantum mechanics and spectroscopy to describe and predict the structure of matter atoms molecules and solids Chapters 14 to 29 present statistical thermodynamics and kinetics and applies their principles to understanding equilibria chemical transformations macromolecular properties and supramolecular machines Each chapter of the book begins with a simplified view of a topic and evolves to more rigorous description in order to provide the student and instructor flexibility to choose the level of rigor and detail that suits them best The textbook treats important new directions in physical chemistry research including chapters on macromolecules principles of interfaces and films for organizing matter and supramolecular machines as well as including discussions of modern nanoscience spectroscopy and reaction dynamics throughout the text

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