

Unit 2 Worksheet 3 – PVTn Problems

On each of the problems below, start with the given P, V, T, or n; then make a decision as to how a change in P, V, T, or n will affect the starting quantity, and then multiply by the appropriate factor. Draw particle diagrams of the initial and final conditions.

1. A sample of gas occupies 150 mL at 25 °C. What is its volume when the temperature is increased to 50 °C? (P and n = constant)

	P	T	V	n
Initial	—	298K	150mL	—
Final	—	323K	?	—
Effect	—	↑	↑	—

$$V_2 = 150\text{mL} \times \frac{323\text{K}}{298\text{K}} = 163\text{mL} \rightarrow 160\text{mL}$$

2. The pressure in a bicycle tire is 105 psi at 25 °C here in Fresno. You take the bicycle up to Huntington, where the temperature is – 5 °C. What is the pressure in the tire? (V and n = constant)

	P	T	V	n
Initial	105psi	298K	—	—
Final	?	268K	—	—
Effect	↓	↓	—	—

$$P_2 = 105\text{psi} \times \frac{268\text{K}}{298\text{K}} = 94.4\text{psi}$$

3. What would be the new pressure if 250 cm³ of gas at standard pressure is compressed to a volume of 150 cm³ ? (T, n = constant)

	P	T	V	n
Initial	1atm	—	250cm ³	—
Final	?	—	150cm ³	—
Effect	↑	—	↓	—

$$P_2 = 1\text{atm} \times \frac{250\text{cm}^3}{150\text{cm}^3} = 1.7\text{atm}$$

or 1300mmHg
or 170kPa
or 25psi

Chemistry Unit 2 Pvtn Problems Answers

JA Banks



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Table of Contents Chemistry Unit 2 PvtN Problems Answers

1. Understanding the eBook Chemistry Unit 2 PvtN Problems Answers
 - The Rise of Digital Reading Chemistry Unit 2 PvtN Problems Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Unit 2 PvtN Problems Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Unit 2 PvtN Problems Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Unit 2 PvtN Problems Answers
 - Personalized Recommendations
 - Chemistry Unit 2 PvtN Problems Answers User Reviews and Ratings
 - Chemistry Unit 2 PvtN Problems Answers and Bestseller Lists
5. Accessing Chemistry Unit 2 PvtN Problems Answers Free and Paid eBooks

- Chemistry Unit 2 Pvtn Problems Answers Public Domain eBooks
 - Chemistry Unit 2 Pvtn Problems Answers eBook Subscription Services
 - Chemistry Unit 2 Pvtn Problems Answers Budget-Friendly Options
6. Navigating Chemistry Unit 2 Pvtn Problems Answers eBook Formats
- ePub, PDF, MOBI, and More
 - Chemistry Unit 2 Pvtn Problems Answers Compatibility with Devices
 - Chemistry Unit 2 Pvtn Problems Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Chemistry Unit 2 Pvtn Problems Answers
 - Highlighting and Note-Taking Chemistry Unit 2 Pvtn Problems Answers
 - Interactive Elements Chemistry Unit 2 Pvtn Problems Answers
8. Staying Engaged with Chemistry Unit 2 Pvtn Problems Answers
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Unit 2 Pvtn Problems Answers
9. Balancing eBooks and Physical Books Chemistry Unit 2 Pvtn Problems Answers
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Unit 2 Pvtn Problems Answers
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemistry Unit 2 Pvtn Problems Answers
- Setting Reading Goals Chemistry Unit 2 Pvtn Problems Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemistry Unit 2 Pvtn Problems Answers
- Fact-Checking eBook Content of Chemistry Unit 2 Pvtn Problems Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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