Using the first molecule as a guide, in Model 12.1 write each molecule's:

#### Model 12.1: Constitutional isomers

Constitutional (also known as structural) isomers have the same chemical formula but different atom connectivity. The following hydrocarbons are all constitutional isomers:

Line structure	Name	Functional group	Chemical formula
	1-butene	Alkene	$C_4H_8$
_			
<u>`</u>			
人			

- a. Name 2-butene 2-butene 2-methyl-1-propene Cyclobutane 1-methylcylopropane
- b. Functional group, and: Alkene Alkene Cycloalkane Cycloalkane
- c. Chemical formula
  C<sub>4</sub>H<sub>8</sub>
  C<sub>4</sub>H<sub>8</sub>
  C<sub>4</sub>H<sub>8</sub>
  C<sub>4</sub>H<sub>8</sub>
  C<sub>4</sub>H<sub>8</sub>
- Based on the information determined in Question 1, which features are sometimes different for each of the molecules above?

The name, shape and functional group

# **Chemistry Week 12 Benchmark Study Guide Answers**

**Liying Dong** 

**Chemistry Week 12 Benchmark Study Guide Answers:** 

### Chemistry Week 12 Benchmark Study Guide Answers Book Review: Unveiling the Power of Words

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