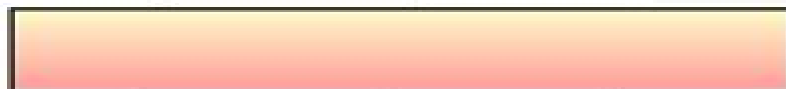


ChemQuest 22



Name: _____

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Information: Terminology

Recall that an ionic bond results from the combination of a metal and a nonmetal. A covalent bond is the type of bond between two nonmetals. Covalent bonds are formed by neutral atoms that share electrons rather than by charged ions. When a compound is formed by sharing electrons, the compound is called a molecule or molecular compound. It is important to note that ionic compounds are not called molecules. The largest class of molecules are called organic molecules. Carbon is the distinguishing mark of organic compounds.

Critical Thinking Questions

1. Circle any of the following compounds that would properly be called a "molecule".



Information: Naming Covalent Compounds

There are several prefixes used to name molecules. The name "carbon oxide" is not sufficient because carbon and oxygen sometimes form CO_2 and sometimes CO . Prefixes are necessary to distinguish between them.

Formula	Name
N_2O_4	dinitrogen tetraoxide
SF_6	sulfur hexafluoride
XeCl_5	xenon pentachloride
SO_3	sulfur trioxide
CO	carbon monoxide

Critical Thinking Questions

2. Fill in the table to indicate which prefix is used to represent the numbers. The first one is done for you.

Number	Prefix
1	mono
2	di
3	tri
4	tetra
5	penta
6	hexa

Chemquest 22 Answer Key

Xiaolong Qi

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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