

Chemistry Worksheet Class 11 on Chapter 1 Some Basic Concepts of Chemistry - Set 3

Q1. In a hydrocarbon, the mass ratio of hydrogen and carbon is 1:3; the empirical formula of hydrocarbon is ____.

- a.) CH_4
- b.) CH_2
- c.) C_2H
- d.) CH_3

Q2. 80g of oxygen contains as many atoms as in ____.

- a.) 10 g of hydrogen
- b.) 5 g of hydrogen
- c.) 8 g of hydrogen
- d.) 1 g of hydrogen

Q3. A mixture of gases contains H_2 and O_2 gases in the ratio 1:4. What is the molar ratio of the two gases in the mixture?

- a.) 16:1
- b.) 2:1
- c.) 1:4
- d.) 4:1

Q4. Which of the following concentration terms is/are independent of temperature?

- a.) Molality
- b.) Mole fraction
- c.) Molarity
- d.) Normality

Q5. Which of the following contains the highest number of atoms?

- a.) 1.0 g of water
- b.) 1.0 g of silver
- c.) 1.0 g of nitrogen
- d.) 1.0 g of propane

Chemistry Chapter 11 Guided Reading Answers

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 $\text{HCl(aq)} + \text{NaOH(aq)} \rightarrow \text{NaCl(aq)} + \text{H}_2\text{O(l)}$
 one mole of HCl would be fully neutralized by one mole of NaOH

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