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corresponding changes in temperature and phase of the substance is observed this is graphically represented by heating and cooling curves

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web determine the final temperature and state of the ammonia once the system reaches equilibrium assuming that no heat is lost to the surroundings use the information below normal boiling point of NH_3 33.3 °C ΔH_{vap} 23.4 kJ/mol specific heat NH_3 1.474 J/g·K specific heat NH_3 2.06 J/g·K

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web once all of the material has solidified at the time indicated by point c the cooling will continue at a rate determined by the heat capacities of the two solids a and b the composition and of course the geometry of the experimental set up by measuring cooling curves for samples of varying composition one can map the entire phase

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