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Sabine Zange



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oxidation $\text{Co} \rightarrow \text{Co}^{2+} + 2\text{e}^-$ reduction $\text{Fe}^{2+} + 2\text{e}^- \rightarrow \text{Fe}$ 24 3 $\text{Ag} \rightarrow \text{Ag}^{+} + \text{e}^-$ 3 $\text{Ag}^{+} + \text{e}^- \rightarrow \text{Ag}$ oxidation reduction 25 $\text{Cu} \rightarrow \text{Cu}^{2+} + 2\text{e}^-$ $\text{Pb} \rightarrow \text{Pb}^{2+} + 2\text{e}^-$ $\text{Cu}^{2+} + 2\text{e}^- \rightarrow \text{Cu}$ oxidation reduction 26 $\text{O}_2 + 4\text{e}^- + 4\text{H}^{+} \rightarrow 2\text{H}_2\text{O}$

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