

Chemistry Study Guide

Periodic Table of the Elements

1 H 1.008																	2 He 4.003																															
3 Li 6.941	4 Be 9.012																	5 B 10.81	6 C 12.01	7 N 14.01	8 O 15.99	9 F 18.99	10 Ne 20.18																									
11 Na 22.99	12 Mg 24.31																	13 Al 26.98	14 Si 28.09	15 P 30.97	16 S 32.06	17 Cl 35.45	18 Ar 39.95																									
19 K 39.09	20 Ca 40.08	21 Sc 44.96	22 Ti 47.88	23 V 50.94	24 Cr 51.99	25 Mn 54.94	26 Fe 55.85	27 Co 58.93	28 Ni 58.71	29 Cu 63.55	30 Zn 65.38	31 Ga 69.72	32 Ge 72.64	33 As 74.92	34 Se 78.96	35 Br 79.90	36 Kr 83.80																															
37 Rb 85.47	38 Sr 87.62	39 Y 88.91	40 Zr 91.22	41 Nb 92.91	42 Mo 95.94	43 Tc 98.91	44 Ru 101.07	45 Rh 102.91	46 Pd 106.42	47 Ag 107.87	48 Cd 112.41	49 In 114.82	50 Sn 118.71	51 Sb 121.76	52 Te 127.60	53 I 126.91	54 Xe 131.29																															
55 Cs 132.91	56 Ba 137.33			57 Hf 178.49	58 Ta 180.95	59 W 183.84	60 Re 186.21	61 Os 190.23	62 Ir 192.22	63 Pt 195.08	64 Au 196.97	65 Hg 200.59	66 Tl 204.38	67 Pb 207.2	68 Bi 208.98	69 Po 209	70 At 210	71 Rn 222																														
87 Fr 223	88 Ra 226			89 Rf 261	90 Db 262	91 Sg 266	92 Bh 264	93 Hs 277	94 Mt 268	95 Ds 271	96 Rg 272	97 Cn 285	98 Uut 289	99 Fl 288	100 Uup 289	101 Lv 293	102 Uus 294	103 Uuo 294																														
		Lanthanide Series <table> <tr> <td>La</td><td>Ce</td><td>Pr</td><td>Nd</td><td>Pm</td><td>Sm</td><td>Eu</td><td>Gd</td><td>Tb</td><td>Dy</td><td>Ho</td><td>Er</td><td>Tm</td><td>Yb</td><td>Lu</td> </tr> <tr> <td>Ac</td><td>Th</td><td>Pa</td><td>U</td><td>Np</td><td>Pu</td><td>Am</td><td>Cm</td><td>Bk</td><td>Cf</td><td>Es</td><td>Fm</td><td>Md</td><td>No</td><td>Lr</td> </tr> </table>																	La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu	Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr
La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu																																		
Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr																																		

Objective:

To understand the *structure of an atom*, and the *nature + properties* of subatomic particles 🎯

Matter:

Anything that has **mass** and **occupies space** ✖️🌐

Examples: air, gases, liquids, solids!

Atoms:

Atoms are the **smallest, indivisible particle** 🧪👉

- **Outer part** → *shell/orbit/energy level* → carries the electrons (e^-)
- **Inner part** → *nucleus* → houses the protons (p^+) and neutrons (n)
- Outer & Inner part → together these are the **subatomic particles**

Subatomic Particle Table:

Particle	Symbol	Mass (amu)	Charge	Location
Proton	p^+	1	+1	Nucleus
Neutron	n	1	0	Nucleus
Electron	e^-	~ 0	-1	Orbit/Shell

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Table of Contents Chemistry Study Guide Review Answers

1. Understanding the eBook Chemistry Study Guide Review Answers
 - The Rise of Digital Reading Chemistry Study Guide Review Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Study Guide Review Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Study Guide Review Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Study Guide Review Answers
 - Personalized Recommendations
 - Chemistry Study Guide Review Answers User Reviews and Ratings
 - Chemistry Study Guide Review Answers and Bestseller Lists

5. Accessing Chemistry Study Guide Review Answers Free and Paid eBooks
 - Chemistry Study Guide Review Answers Public Domain eBooks
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 - Chemistry Study Guide Review Answers Compatibility with Devices
 - Chemistry Study Guide Review Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry Study Guide Review Answers
 - Highlighting and Note-Taking Chemistry Study Guide Review Answers
 - Interactive Elements Chemistry Study Guide Review Answers
8. Staying Engaged with Chemistry Study Guide Review Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Study Guide Review Answers
9. Balancing eBooks and Physical Books Chemistry Study Guide Review Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Study Guide Review Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemistry Study Guide Review Answers
 - Setting Reading Goals Chemistry Study Guide Review Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemistry Study Guide Review Answers
 - Fact-Checking eBook Content of Chemistry Study Guide Review Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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