

Phase Change Worksheet

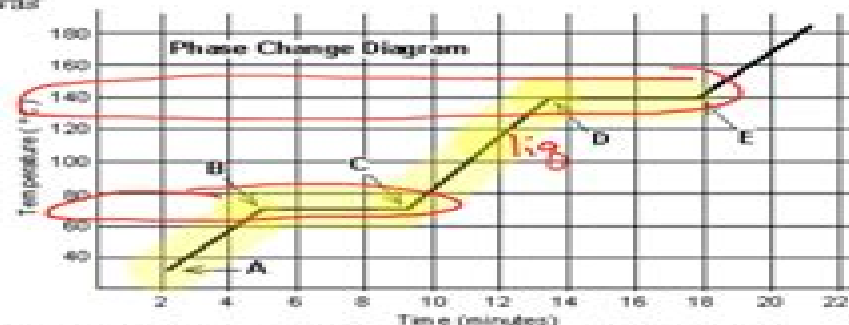
Name: _____

Date: _____ Period: _____

The graph was drawn from data collected as a substance was heated at a constant rate.

Use the graph to answer the following questions 1-23.

freeze evaporate
Condensation
Melting deposition
Sublimation



1. - 22. At point A, the beginning of observations, the substance exists in a solid state. Material in this phase has (1) definite volume and (2) definite shape. With each passing minute, (3) heat is added to the substance. This causes the molecules of the substance to (4) move more rapidly which we detect by a (5) temperature rise in the substance. At point B, the temperature of the substance is (6) 70 °C. The solid begins to (7) Melt. At point C, the substance is completely (8) Melted or in a (9) liquid state. Material in this phase has (10) definite volume and (11) indefinite shape. The energy put to the substance between minutes 5 and 9 was used to convert the substance from a (12) Solid to a (13) liquid.

Between 9 and 13 minutes, the added energy increases the (14) temp of the substance. During the time from point D to point E, the liquid is (15) evaporating. By point E, the substance is completely in the (16) gas phase. Material in this phase has (17) indef volume and (18) indef shape. The energy put to the substance between minutes 13 and 18 converted the substance from a (19) lig to a (20) gas state. Beyond point E, the substance is still in the (21) gas phase, but the molecules are moving (22) faster as indicated by the increasing temperature.

23. Which of these three substances was likely used in this phase change experiment?

Substance	Melting point	Boiling point
<u>Bolognium</u>	<u>30 °C</u>	<u>100 °C</u>
<u>Unobtainium</u>	<u>40 °C</u>	<u>140 °C</u>
<u>Eccsium</u>	<u>70 °C</u>	<u>140 °C</u>

Chemistry Study Guide Phase Change Answer Questions

Jill Meryl Levy



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