

1. Using the first molecule as a guide, in Model 12.1 write each molecule's:

### Model 12.1: Constitutional isomers

Constitutional (also known as structural) isomers have the same chemical formula but different atom connectivity. The following hydrocarbons are all constitutional isomers:

| Line structure                                                                    | Name     | Functional group | Chemical formula |
|-----------------------------------------------------------------------------------|----------|------------------|------------------|
|  | 1-butene | Alkene           | $C_4H_8$         |
|  |          |                  |                  |
|  |          |                  |                  |
|  |          |                  |                  |
|  |          |                  |                  |
|  |          |                  |                  |

a. Name

2-butene  
2-butene  
2-methyl-1-propene  
Cyclobutane  
1-methylcyclopropane

b. Functional group, and:

Alkene  
Alkene  
Alkene  
Cycloalkane  
Cycloalkane

c. Chemical formula

$C_4H_8$   
 $C_4H_8$   
 $C_4H_8$   
 $C_4H_8$   
 $C_4H_8$

2. Based on the information determined in Question 1, which features are sometimes different for each of the molecules above?

The name, shape and functional group

# Chemistry Week 12 Benchmark Study Guide Answers

**RM Cervero**



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