

Chemistry Study Guide

Periodic Table of the Elements

1 H 1.008																	2 He 4.003																															
3 Li 6.941	4 Be 9.012											5 B 10.81	6 C 12.01	7 N 14.01	8 O 16.00	9 F 18.998	10 Ne 20.18																															
11 Na 22.99	12 Mg 24.31											13 Al 26.98	14 Si 28.09	15 P 30.97	16 S 32.07	17 Cl 35.45	18 Ar 39.95																															
19 K 39.10	20 Ca 40.08	21 Sc 44.96	22 Ti 47.88	23 V 50.94	24 Cr 52.00	25 Mn 54.94	26 Fe 55.85	27 Co 58.93	28 Ni 58.71	29 Cu 63.55	30 Zn 65.38	31 Ga 69.72	32 Ge 72.64	33 As 74.92	34 Se 78.96	35 Br 79.90	36 Kr 83.80																															
37 Rb 85.47	38 Sr 87.62	39 Y 88.91	40 Zr 91.22	41 Nb 92.91	42 Mo 95.94	43 Tc 98.91	44 Ru 101.07	45 Rh 102.91	46 Pd 106.42	47 Ag 107.87	48 Cd 112.41	49 In 114.82	50 Sn 118.71	51 Sb 121.76	52 Te 127.60	53 I 126.91	54 Xe 131.29																															
55 Cs 132.91	56 Ba 137.33			57 Hf 178.49	58 Ta 180.95	59 W 183.84	60 Re 186.21	61 Os 190.23	62 Ir 192.22	63 Pt 195.08	64 Au 196.97	65 Hg 200.59	66 Tl 204.38	67 Pb 207.2	68 Bi 208.98	69 Po 209	70 At 210	71 Rn 222																														
87 Fr 223	88 Ra 226			89 Rf 261	90 Db 262	91 Sg 266	92 Bh 264	93 Hs 277	94 Mt 268	95 Ds 271	96 Rg 272	97 Cn 285	98 Uut 289	99 Fl 288	100 Uup 289	101 Lv 293	102 Uus 294	103 Uuo 294																														
Lanthanide Series <table> <tr> <td>La 138.91</td> <td>Ce 140.12</td> <td>Pr 140.91</td> <td>Nd 144.24</td> <td>Pm 145</td> <td>Sm 150.36</td> <td>Eu 151.96</td> <td>Gd 157.25</td> <td>Tb 158.93</td> <td>Dy 162.50</td> <td>Ho 164.93</td> <td>Er 167.26</td> <td>Tm 168.93</td> <td>Yb 173.05</td> <td>Lu 174.97</td> </tr> <tr> <td>Ac 227</td> <td>Th 232.04</td> <td>Pa 231</td> <td>U 238.03</td> <td>Np 237</td> <td>Pu 244</td> <td>Am 243</td> <td>Cm 247</td> <td>Bk 247</td> <td>Cf 251</td> <td>Es 252</td> <td>Fm 257</td> <td>Md 258</td> <td>No 259</td> <td>Lr 262</td> </tr> </table>																			La 138.91	Ce 140.12	Pr 140.91	Nd 144.24	Pm 145	Sm 150.36	Eu 151.96	Gd 157.25	Tb 158.93	Dy 162.50	Ho 164.93	Er 167.26	Tm 168.93	Yb 173.05	Lu 174.97	Ac 227	Th 232.04	Pa 231	U 238.03	Np 237	Pu 244	Am 243	Cm 247	Bk 247	Cf 251	Es 252	Fm 257	Md 258	No 259	Lr 262
La 138.91	Ce 140.12	Pr 140.91	Nd 144.24	Pm 145	Sm 150.36	Eu 151.96	Gd 157.25	Tb 158.93	Dy 162.50	Ho 164.93	Er 167.26	Tm 168.93	Yb 173.05	Lu 174.97																																		
Ac 227	Th 232.04	Pa 231	U 238.03	Np 237	Pu 244	Am 243	Cm 247	Bk 247	Cf 251	Es 252	Fm 257	Md 258	No 259	Lr 262																																		

Objective:

To understand the *structure of an atom*, and the *nature + properties* of subatomic particles ✎

Matter:

Anything that has **mass** and **occupies space** ✎

Examples: air, gases, liquids, solids!

Atoms:

Atoms are the **smallest, indivisible particle** ✎

- **Outer part** → *shell/orbit/energy level* → carries the electrons (e^-)
- **Inner part** → *nucleus* → houses the protons (p^+) and neutrons (n)
- Outer & Inner part → together these are the **subatomic particles**

Subatomic Particle Table:

Particle	Symbol	Mass (amu)	Charge	Location
Proton	p^+	1	+1	Nucleus
Neutron	n	1	0	Nucleus
Electron	e^-	~ 0	-1	Orbit/Shell

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Enrique Olivas



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Table of Contents Chemistry Study Guide Pearson

1. Understanding the eBook Chemistry Study Guide Pearson
 - The Rise of Digital Reading Chemistry Study Guide Pearson
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Study Guide Pearson
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Study Guide Pearson
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Study Guide Pearson
 - Personalized Recommendations
 - Chemistry Study Guide Pearson User Reviews and Ratings

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 - Participating in Virtual Book Clubs
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 - Creating a Diverse Reading Collection Chemistry Study Guide Pearson
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 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chemistry Study Guide Pearson
 - Setting Reading Goals Chemistry Study Guide Pearson
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemistry Study Guide Pearson
 - Fact-Checking eBook Content of Chemistry Study Guide Pearson
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
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 - Interactive and Gamified eBooks

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