

12

STOICHIOMETRY

Practice Problems

In your notebook, solve the following problems.

SECTION 12.1 THE ARITHMETIC OF EQUATIONS

Use the 3-step problem-solving approach you learned in Chapter 1.

1. An apple pie needs 10 large apples, 2 cups (top and bottom), and 1 tablespoon of cinnamon. Write a balanced equation that fits this situation. How many apples are needed to make 20 pies?
2. Two moles of potassium chloride and three moles of oxygen are produced from the decomposition of two moles of potassium chlorate, $2\text{KClO}_3(\text{s})$. Write the balanced equation. How many moles of oxygen are produced from 12 moles of potassium chlorate?
3. Using the equation from problem 2, how many moles of oxygen are produced from 14 moles of potassium chlorate?
4. Two molecules of hydrogen react with one molecule of oxygen to produce two molecules of water. How many molecules of water are produced from 3.8×10^{23} molecules of oxygen? How many moles of water are produced from 22.5 moles of oxygen?

SECTION 12.2 CHEMICAL CALCULATIONS

1. Calculate the number of moles of hydrogen chloride produced from 10 moles of hydrogen.



2. Calculate the number of moles of chlorine needed to form 14 moles of iron(II) chloride.



3. Calculate the number of grams of nitrogen dioxide that are produced from 4 moles of nitric oxide.



4. Calculate the mass of oxygen produced from the decomposition of 75.0 g of potassium chlorate.



5. Calculate the mass of silver needed to react with chlorine to produce 94 g of silver chloride. Write a balanced equation first.

6. How many liters of carbon monoxide at STP are needed to react with 4.00 g of oxygen gas to produce carbon dioxide?



7. Calculate the number of liters of oxygen gas needed to produce 15.0 liters of dinitrogen monoxide. Assume all gases are at the same conditions of temperature and pressure.



Chapter 12 Chemistry Stoichiometry Study Guide Answers

Hussin A.Rothana



Chapter 12 Chemistry Stoichiometry Study Guide Answers:

This book delves into Chapter 12 Chemistry Stoichiometry Study Guide Answers. Chapter 12 Chemistry Stoichiometry Study Guide Answers is a vital topic that must be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Chapter 12 Chemistry Stoichiometry Study Guide Answers, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:
 - Chapter 1: Introduction to Chapter 12 Chemistry Stoichiometry Study Guide Answers
 - Chapter 2: Essential Elements of Chapter 12 Chemistry Stoichiometry Study Guide Answers
 - Chapter 3: Chapter 12 Chemistry Stoichiometry Study Guide Answers in Everyday Life
 - Chapter 4: Chapter 12 Chemistry Stoichiometry Study Guide Answers in Specific Contexts
 - Chapter 5: Conclusion
 2. In chapter 1, this book will provide an overview of Chapter 12 Chemistry Stoichiometry Study Guide Answers. This chapter will explore what Chapter 12 Chemistry Stoichiometry Study Guide Answers is, why Chapter 12 Chemistry Stoichiometry Study Guide Answers is vital, and how to effectively learn about Chapter 12 Chemistry Stoichiometry Study Guide Answers.
 3. In chapter 2, the author will delve into the foundational concepts of Chapter 12 Chemistry Stoichiometry Study Guide Answers. The second chapter will elucidate the essential principles that must be understood to grasp Chapter 12 Chemistry Stoichiometry Study Guide Answers in its entirety.
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 5. In chapter 4, the author will scrutinize the relevance of Chapter 12 Chemistry Stoichiometry Study Guide Answers in specific contexts. This chapter will explore how Chapter 12 Chemistry Stoichiometry Study Guide Answers is applied in specialized fields, such as education, business, and technology.
 6. In chapter 5, this book will draw a conclusion about Chapter 12 Chemistry Stoichiometry Study Guide Answers. The final chapter will summarize the key points that have been discussed throughout the book.
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