

Characteristics of Chemical Equilibrium

The purpose of this lab is to investigate how changing various factors affect an equilibrium system. This lab consists of four independent activities. Each activity changes a component (concentration, temperature, and pressure) in an equilibrium system. Students observe how the changes affect the equilibrium system and analyze the changes using Le Chatelier's Principle.

Please explain your answers to the following questions using Le Chatelier's Principle.

Activity A: Nature of Equilibrium



Colorless

Pink



I

If HCl solution is added to the solution in equilibrium A, would the solution become more pink or less pink?

If NaOH solution is added to the solution, which direction would the equilibrium shift to? Would the color become more pink or less pink when NaOH solution is added to the equilibrium?

Characteristics Of Chemical Equilibrium Lab Answers

N Colangelo



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