

REVIEW FOR THERMOCHEMISTRY TEST

1. The combustion of methanol is shown by the following equation:



- (a) Using the data below, find the heat of reaction for the combustion of methanol.



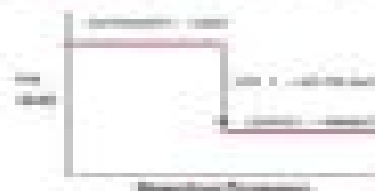
- (b) What is the molar enthalpy of combustion for methanol?

$$\Delta H = -1278 \text{ kJ/mol} \text{ and } -639 \text{ kJ/mol}$$

- (c) Express the energy change for the target reaction in the form $\Delta H_{\text{reaction}}$.

$$\Delta H_{\text{reaction}} \text{ CH}_3\text{OH} = -639 \text{ kJ/mol}$$

- (d) Draw the potential energy diagram to illustrate the energy change for the target reaction.



Chemistry Thermochemistry Unit Test Review

Léo Pomar



Chemistry Thermochemistry Unit Test Review:

Embracing the Song of Appearance: An Psychological Symphony within **Chemistry Thermochemistry Unit Test Review**

In a world taken by screens and the ceaseless chatter of instant transmission, the melodic elegance and psychological symphony created by the prepared word usually disappear into the backdrop, eclipsed by the constant sound and interruptions that permeate our lives. Nevertheless, situated within the pages of **Chemistry Thermochemistry Unit Test Review** a charming literary value overflowing with raw feelings, lies an immersive symphony waiting to be embraced. Constructed by an outstanding musician of language, that captivating masterpiece conducts viewers on a mental journey, skillfully unraveling the concealed songs and profound affect resonating within each carefully constructed phrase. Within the depths of this moving analysis, we shall explore the book is central harmonies, analyze its enthralling publishing fashion, and submit ourselves to the profound resonance that echoes in the depths of readers souls.

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