# CHEMISTRY 1 FINAL EXAM REVIEW

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Amounts.

Name Section Date

#### Chemistry Review: Writing and Balancing Equations

Directions: Solve each problem and answer each question

- 1. Balance each equation below by adding the correct coefficients
  - a. Goth, + Note + Noti + Goth,
  - B. \_\_HCN+\_MgGOH;-b\_MgGNG;+\_H,O
  - c \_H90,+\_0910,50,-9\_NH,NO,+\_H,50,
  - d. \_2MDDL+\_ALO, +\_2MO+\_AMDDL
  - a. \_\_C,H\_a + \_\_O, -> \_\_OO, + \_\_H,O
- Write and balance each equation, include phase labels:
  - Aguerous sodium hydroxide reacts with aquerous suffuric acid to form aquerous sodium suffare and reader
  - Solid into (20) hydroxide and aqueous sodium carbonate are formed from the reaction of solid into (20) carbonate and aqueous sodium hydroxide
  - Solid aluminum reacts with aqueous hydrobromic acid to form aqueous aluminum bromide and hydrogen gas
  - Gazeous ethane (dicarbon hesshydride) reacts with oxygen gas to form passous carbon-dicaids and passous water
  - Solid calcium phosphite reacts with aqueous copper(10) nitrate to form aqueous calcium nitrate and solid copper (10) phosphite

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## 21 CHEMISTRY WORKSHEETS WORKED ANSWER KEY

### **Cp Chemistry Final Exam Review Topics**

**Roman Wölfel** 

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