AP Chemistry Semester 1 Exam Study Guide

Multiple Choice

Identify the choice that best completes the statement or answers the question

 $HNO_3(aq) + CaCO_3(s) \rightarrow Ca(NO_3)_2(aq) + CO_2(g) + H_2O(l)$

- a. 1
- 5. 2.
- c. 3
- 6. 5
- e. 4

Predict the product in the combination reaction below.

Al (s) + N₂ (g) \rightarrow

- a. AliN
- b. AlaN
- c. AlNa
- d. Al₃N₂
- e. AlNa

What is the empirical formula of a compound that contains 27.0% S, 13.4% O, and 59.6% Cl by mass?

- a. SOCI
- b. SOCIa
- c. SyOCE
- d. SO₂C1
- e. CISO+

4. A compound is composed of only C, H, and O. The combustion of a 0.519-g sample of the compound yields

1.24 g of CO2 and 0.255 g of H2O. What is the empirical formula of the compound?

- a. C₆H₆O
- b. C.H.O.
- c. CH₂O
- d. C₂H₆O₃
- e. C2H6O2

Propane (C₃H₆) reacts with oxygen in the air to produce carbon dioxide and water. In a particular experiment,
 38.0 grams of carbon dioxide are produced from the reaction of 22.05 grams of propane with excess oxygen.

What is the % yield in this reaction?

- a. 38.0
- 6. 57.6

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Robert R. Redfield

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