

Lewis Structures & Shapes

Part A: Follow the instructions in the procedure to complete this table.

Formula	a. No. of valence electrons	b. Lewis Structure	c. Shape/ Molecular geometry	d. Bond Angles	e. Dipole Moment	f. Initials (if required)
H ₂ O (example)	8		bent	~109.5°	☒ Yes ☐ No	
CH ₃ Cl	14		tetrahedral	109.5°	☒ Yes ☐ No	
NH ₃	8		trigonal pyramidal	109.5°	☒ Yes ☐ No	
H ₂ S	8		bent/V-shaped	109.5°	☒ Yes ☐ No	
CH ₂ O	12		trigonal planar	120°	☒ Yes ☐ No	

Chemfiesta Answers Lewis Dot Structures

AN Whitehead



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