

11.1 - Describing Chemical Reactions

1. What two substances reacted in the Hindenburg crash?
2. In general, what happens in a **chemical reaction**?
3. Look at Figure 11.1. What are some of the evidences of **chemical change** (a.k.a. chemical reaction)?
4. Give an example of a **word equation** for a chemical reaction (you can make one up or choose one from the book).
5. How is a **chemical equation** different from a word equation?
6. What is missing in a skeleton equation?
7. What symbols are used to represent the phases of substances in chemical reactions?
8. What does the symbol (aq) mean? _____
9. Use the space below to do Practice Problems #1 and #2.
1) _____ 2) _____
10. A complete description of a reaction must include _____,
but also _____.
11. Define the following:
 - a. **coefficients** -
 - b. **subscripts** -
 - c. **balanced equation** -
12. How does the **law of conservation of mass** relate to balancing reaction equations?
13. In a chemical reaction, what **MUST** be the same for the reactants and the products?
(There are two great answers to this question!)

Chapter 11 Guided Notes Name 11 1 A Describing Chemical

ML Morrison



Chapter 11 Guided Notes Name 11 1 A Describing Chemical:

Unveiling the Magic of Words: A Report on "**Chapter 11 Guided Notes Name 11 1 A Describing Chemical**"

In a global defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their ability to kindle emotions, provoke contemplation, and ignite transformative change is really awe-inspiring. Enter the realm of "**Chapter 11 Guided Notes Name 11 1 A Describing Chemical**," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve in to the book is central themes, examine its distinctive writing style, and assess its profound affect the souls of its readers.

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