

ACIDS and BASES: Multiple Choice

1. Apples have a pH of 3.9, beans have a pH of 5.9, and coconuts a pH of 7.9.

Which of the following statements is true?

- A) Beans are 100 times more acidic than coconut, and 10 000 times more acidic than apples.
- B) Apples are 100 times more acidic than beans, and 10 000 times more acidic than coconut.
- C) Beans are 100 times more acidic than apples, and 10 000 times more acidic than coconut.
- D) Coconut is 100 times more acidic than beans, and 10 000 times more acidic than apples.

ANS: B

1. Which of the following is NOT a strong base?

- A) $\text{NaOH}_{(aq)}$
- B) $\text{KOH}_{(aq)}$
- C) $\text{LiOH}_{(aq)}$
- D) $\text{CaOH}_{(aq)}$
- E) $\text{Mg(OH)}_{2(aq)}$

ANS: D

1. Which compound is a strong base?

- A) B(OH)_3
- B) KOH
- C) SO_4^{2-}
- D) HCO_3^-
- E) NH_3

ANS: B

1. Which of the following indicates an acidic solution?

- A) pH = 7.5
- B) pH = 7
- C) pH = 6.5
- D) more than one correct response
- E) no correct response

ANS: C

1. Which of the following indicates an acidic solution?

- A) pH = 5.5
- B) pH = 7
- C) pH = 6.5
- D) more than one correct response
- E) no correct response

ANS: D

1. Which of the following indicates an acidic solution?

- A) pH = 8.5
- B) pH = 7
- C) pH = 7.5
- D) more than one correct response
- E) no correct response

ANS: E

1. Which of the following equations represents an acid-base neutralization reaction?

- A) $\text{H}_2\text{SO}_4 + \text{Zn} \rightarrow \text{ZnSO}_4 + \text{H}_2$
- B) $\text{HNO}_3 + \text{KOH} \rightarrow \text{KNO}_3 + \text{H}_2\text{O}$
- C) $\text{Ba(OH)}_2 + \text{Na}_2\text{SO}_4 \rightarrow \text{BaSO}_4 + 2\text{NaOH}$
- D) more than one correct response
- E) none of the above

ANS: B

Chemistry Test Acids Bases Multiple Choice

KJ Lindholm-Leary



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