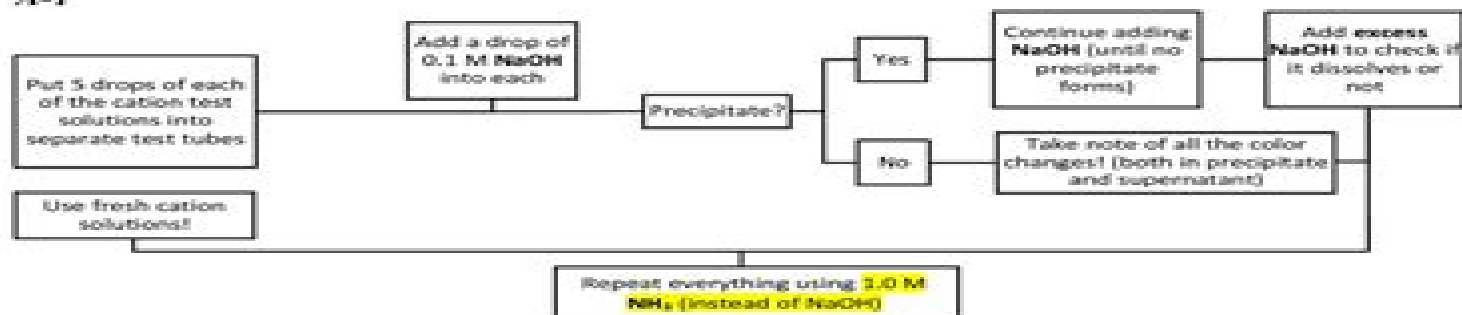


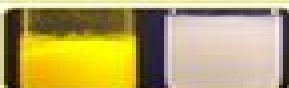
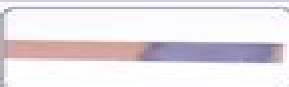
1. CATIONS – Ca^{2+} , Fe^{3+} , Cu^{2+} , Zn^{2+} , NH_4^+

Elimination Tests

A-1



Confirmatory Tests

	Cu^{2+} • Deep blue complex (upon mixing of solution with excess NH_3)
	Fe^{3+} • Blood red complex (FeSCN^{2+}) (upon addition of 2 drops of 0.1 M KSCN) • Dark blue precipitate ($\text{KFe}^{III}\text{Fe}^{II}(\text{CN})_6$) (upon addition of 2 drops of 0.1 M $\text{K}_4\text{Fe}(\text{CN})_6$)
	Ca^{2+} • White precipitate (upon mixing with 2 drops of 0.1 M $(\text{NH}_4)_2\text{C}_2\text{O}_4$)
	Zn^{2+} • Yellow precipitate ($\text{K}_2\text{ZnFe}(\text{CN})_6$) (upon addition of 2 drops of 0.1 M $\text{K}_4\text{Fe}(\text{CN})_6$) • White precipitate ($\text{K}_2\text{Zn}(\text{C}_2\text{O}_4)_2$) (upon addition of 0.1 M $\text{K}_4\text{Fe}(\text{CN})_6$)
	NH_4^+ • Red litmus paper turns blue upon contact with vapor formed by mixing 30 drops of solution with 5 drops of 1.0 M NaOH

Moisten 2 pieces of litmus paper (red and blue) with distilled water

Stick to the underside of watch glass

Immediately cover evaporating dish (containing the solution above) with the watch glass (litmus paper facing the solution)

Do NOT allow solution to touch the litmus paper!

Cations And Ions Lab Report

M Mosston

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the name bar.

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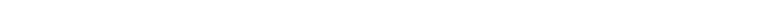
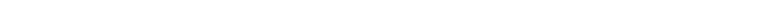
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