# Physical Equilibria – Phases Equilibria

$$H_2O(s) \rightleftharpoons H_2O(l)$$

At 0 °C (273 K) there exists a phase equilibrium between solid and liquid H<sub>2</sub>O.

At 100 °C (373 K) there exists a phase equilibrium between liquid and gaseous H<sub>2</sub>O.

$$H_2O(I) \rightleftharpoons H_2O(g)$$



# **Concept Review Phase Equilibrium**

Allan Blackman, Daniel
Southam, Gwendolyn Lawrie, Natalie
Williamson, Christopher Thompson

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