Chemistry 11: Study Guide for Mole Concept Test Name:			Block: _	
		My "Mole Concept Test" will take place on:!		
Before	Lweite	my "Mole Concept Test", I will be able to:		
	evolui	n why the mole is used in chemistry		
	0	know how to find atomic mass on the periodic table		
		know the relationship between atomic mass and molar mass		
	0	recognize that a mole is a unit for counting atoms, ions, and molecules		
0	perform calculations involving the mole			
	0	determine the molar mass of an element or compound		
	0	convert among particles, moles, and mass		
	under	stand the relationship between moles and gases at STP		
	0	know the volume of a mole of gas at STP (i. e. 22.4 L)		
		convert among moles, mass, and volume of a gas at STP		
0	perform calculations involving molecular and empirical formulae			
	0	distinguish between molecular and empirical formulae		
	0			
	.0			
	.0	determine a molecular formula from the molecular mass and empirical formu	da	
	Describe concentration in terms of molarity			
	(0)	Know that the units for molarity/concentration are mol/L or M		
	0	Write a procedure for how to prepare a solution of known molarity (a standar	rd solution)	
	Perfor	Perform calculations involving molarity		
	-0	Perform calculations relating mass, moles, volume, and molarity		
	Understand the meanings of the following vocabulary words:			
	0	atomic mass		
	0			
	0			
	0	molar mass		
	(0)	molar solution		
	0			
	0	molecular formula		
	0	molecular mass		
	0	quantitative transfer		
	0	percentage composition		
	0	standard solution		
	0	STP		

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Michael Seilmaier

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