



ORGANIC NOMENCLATURE 2

Q	Name	Structural formula	Displayed formula	Skeletal formula
1	2,3-dimethylbutane	$\begin{array}{c} \text{CH}_3\text{CH}(\text{CH}_3)\text{CH}(\text{CH}_3)\text{CH}_3 \\ \text{CH}_3\text{CH}(\text{CH}_3)\text{CH}(\text{CH}_3)\text{CH}_3 \end{array}$	$\begin{array}{c} \text{H} & \text{H} & \text{H} & \text{H} \\ & & & \\ \text{H}-\text{C} & -\text{C} & -\text{C} & -\text{C}-\text{H} \\ & & & \\ \text{H} & \text{H} & \text{H} & \text{H} \end{array}$	
2	4-bromo-2,2-dichloropentane	$\begin{array}{c} \text{CH}_3-\text{CH}(\text{Cl})-\text{CH}(\text{Cl})-\text{CH}_2-\text{CH}_3 \\ \text{CH}_3-\text{CH}(\text{Cl})-\text{CH}(\text{Cl})-\text{CH}_2-\text{CH}_3 \end{array}$	$\begin{array}{c} \text{H} & \text{H} & \text{H} & \text{H} & \text{H} \\ & & & & \\ \text{H}-\text{C} & -\text{C} & -\text{C} & -\text{C} & -\text{C}-\text{H} \\ & & & & \\ \text{H} & \text{Cl} & \text{Cl} & \text{H} & \text{H} \end{array}$	
3	butanone	$\text{CH}_3\text{CH}_2\text{C}(=\text{O})\text{CH}_3$	$\begin{array}{c} \text{H} & \text{H} & \text{H} & \text{H} \\ & & & \\ \text{H}-\text{C} & -\text{C} & -\text{C} & -\text{C}-\text{H} \\ & & & \\ \text{H} & \text{H} & \text{O} & \text{H} \end{array}$	
4	but-2-ene	$\text{CH}_3\text{CH}=\text{CHCH}_3$	$\begin{array}{c} \text{H} & \text{H} & \text{H} & \text{H} \\ & & & \\ \text{H}-\text{C} & -\text{C} & =\text{C} & -\text{C}-\text{H} \\ & & & \\ \text{H} & \text{H} & \text{H} & \text{H} \end{array}$	
5	butan-2-ol	$\text{CH}_3\text{CH}(\text{OH})\text{CH}_2\text{CH}_3$	$\begin{array}{c} \text{H} & \text{H} & \text{H} & \text{H} \\ & & & \\ \text{H}-\text{C} & -\text{C} & -\text{C} & -\text{C}-\text{H} \\ & & & \\ \text{H} & \text{OH} & \text{H} & \text{H} \end{array}$	

Chemsheets Organic Nomenclature

J Ma



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